

22372

3,1560

\$/035/61/000/005/003/042 A001/A101

AUTHORS:

Mustel', E.R., Galkin, L.S.

TITLE:

The spectrometric studying of hydrogen lines in spectra of peculiar stars of class AO. Part I. Hydrogen lines in spectra of "manganese", "silicon" and "magnesium" stars

PERIODICAL:

Referativnyy zhurnal. Astronomiya i Geodeziya, no. 5, 1961, 22, abstract 5A149 ("Izv. Krymsk. astrofiz. observ.", 1960, v. 22, 225-233, Engl. summary)

TEXT: The authors studied the contours of hydrogen lines in spectra of ten peculiar stars of class AO with enhanced lines of manganese, silicon and magnesium. They obtained hydrogen line contours in spectra of eight comparison stars of classes B9 - A1. Equivalent widths of hydrogen lines H β , H γ , H δ , H ϵ H5 and K-line (Ca II) were determined for all stars studied. The contours of hydrogen lines in spectra of "silicon" stars are identical to the contours of corresponding hydrogen lines in the spectra of comparison stars of class AO V. The contours of hydrogen lines in spectra of "manganese" and "magnesium" stars

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S/035/61/000/005/003/042 A001/A101

The spectrometric studying of hydrogen lines ...

studied are shallower and narrower than the corresponding hydrogen lines in the spectra of comparison stars of class AO III. This indicates either the relatively high luminosity of the stars considered or specific conditions in their atmospheres.

From author's summary

[Abstracter's note: Complete translation]

Card 2/2

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S/035/61/000/005/007/042 A001/A101

AUTHORS:

Mustel', E.R., Galkin, L.S.

TITLE:

The spectrophotometric study of hydrogen lines in spectra of peculiar stars of class A. Part II

PERIODICAL:

Referativnyy zhurnal. Astronomiya i Geodeziya, no. 5, 1961, 32, abstract 5A227 ("Izv. Krymsk. astrofiz. observ.", 1960, v. 24, 78-90, Engl. summary)

TEXT: This is the continuation of the authors' study (RZhAstr, 1955, no. 10, 4276; 1956, no. 9, 5047). Balmer absorption lines in spectra of A-class peculiar stars are studied. The authors present lists of peculiar and normal stars, as well as the graphs of line profiles in spectra of peculiar and standard stars. Each profile was plotted from several spectrograms, to increase their accuracy. The results of comparing the profiles of peculiar and standard stars are presented in graphical form; they show that in most cases the profiles of the Balmer series in the spectra of peculiar stars agree well with the corresponding profiles in the spectra of the normal stars of a similar spectral class. Thus, in all these cases, the structure of peculiar stars atmospheres can not apparently differ mark-

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The spectrophotometric study ...

edly from the structure of normal stars atmospheres of a similar class. To confirm additionally this conclusion, electronic pressure $n_e\ (n_m)$ was determined for a number of stars studied by the formula: $\lg n_e = 23.26$ - 7.5 $\lg n_m$; the quantity N_{O2} H was determined by the formula: N_{O2} H = $(mc^2/\pi\,e^2\,\lambda^2\,f)$ W_{\(\lambda\)}, and the quantity $n_e\ (H_{\(\gamma\)}$) by the formula: $w_0^2 = kN_{O2}Hn_e\ (R_c/O.45)^{3/2}$ assuming the value of Ho2H already calculated. All these parameters are presented in tables. Graphs are also presented illustrating the relations between $\lg n_e\ (n_m)$ and $\lg n_e\ (H_{\(\gamma\)}$), and between $\lg N_{O2}H$ and $\lg n_e\ (H_{\(\gamma\)}$). The analysis of these graphs also confirms that apparently the structure of atmospheres of the most peculiar stars differs slightly from the structure of atmospheres of the normal stars of a similar class. It is noted that in some cases (e.g. α Psc) the profiles of Balmer lines of peculiar stars do differ noticeably from the corresponding profiles of the standard stars spectra. There are 9 references.

From authors' summary

[Abstracter's note: Complete translation]

Card 2/2

s/035/62/000/007/024/083

AUTHORS:

Mustel', E. R., Kopylov, I. M., Galkin, L. S., Kumaygorodskaya, R.N.,

Bartash, T. M.

TTTLE:

Spectrophotometric study of Nova Herculis 1960. I.

PERIODICAL: Referativnyy zhurnal, Astronomiya i Geodeziya, no. 7, 1962, 31,

abstract 7A236 ("Izv. Krymsk. astrofiz. observ.", 1961, v. 26,

181 - 216; English summary)

About 120 spectrograms of Nova Herculis and (Aql taken as a standard were taken in March - April 1960 with the 122-cm reflector of the Crimean Astrophysical Observatory, mainly with a quartz spectrograph with dispersion of 155 A/mm at Hγ. The following quantities were determined: equivalent widths Wh and values of $\Delta\lambda$ (km/sec) for emission hydrogen lines H β -H $_{9}$, as well as central intensities Io with respect to continuous spectrum for all identified emission lines in the spectrum of N Her. Changes of these characteristics of emission lines in the course of time were generally analyzed. The average speed of envelope expansion was estimated (1,850 km/sec) from the width of hydrogen lines.

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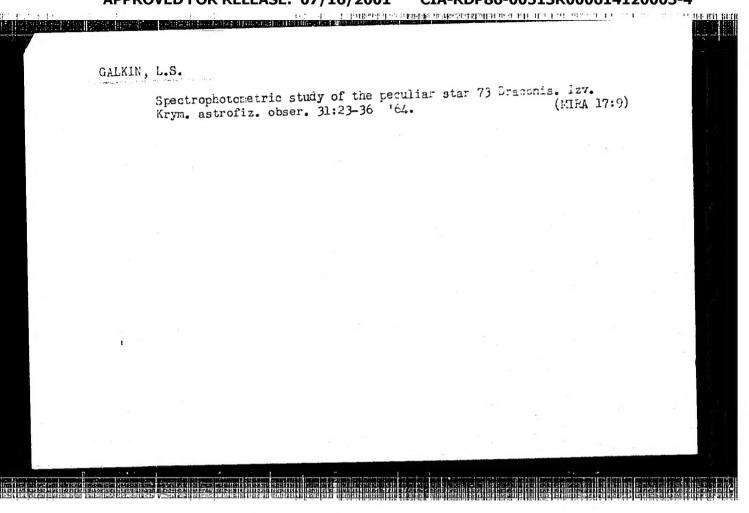
Spectrophotometric study of Nova Herculis 1960. I. A001/A101

Approximate brightness and date of maximum luminosity of N Her were determined from spectral changes observed in this Nova: $m_{max}=2^m50\cdot10^m.17$ (probable error); March $1\pm2^d.5$ (probable error). The absolute magnitude of the Nova in maximum was determined ($M_V=-10^m.0$), as well as distance to the star (R=1,250 pc) and total interstellar absorption in visual light at star distance ($A_V=2.0^m.0$). By comparing with ξ Aql relative energy distribution was obtained for each night in continuous spectrum of N Her within the wavelength range $\lambda \lambda 3,512-5,050$. Balmer decrement was calculated from lines $H\beta$ - H_Q . There are 12 references.

From authors' summary

[Abstracter's note: Complete translation]

Card 2/2



SHUBIN, I., (Sverdlovsk); LIFOROV, G., (ROSTOV-Na-DONU); PARUSHAVICHUS, G., (Vil'nyus); GALKIN, M., (Alma-Ata); KASHTAN'YER, Al.; ANATOL'YEV, E.; SERGEYEV, N.; VASIL'YEV, K.

News from everywhere. Sov.foto 21 no.3:44-46 Mr '61.

(MIRA 14:4)

1. Predsedatel' fotosektsii Soyuza zhurnalistov (for Galkin).

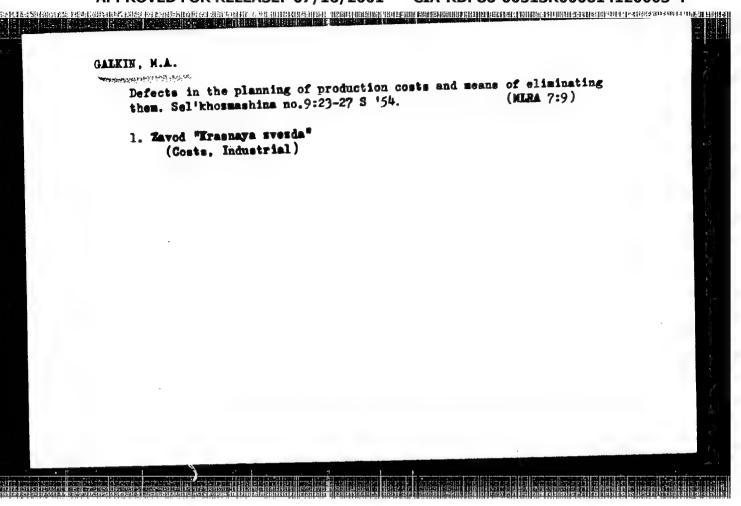
(Photography)

GALKIN, M.

Power of observation and diligence. Sov. foto 22 no.12:28-29
(MIRA 16:1)

1. Predsedatel' fotosektsii Soyusa zhurnalistov KazSSR.

(Photegraphers, Russian)



CAIKIN Mikhail Aleksandroyich; NIKITIN, Viktor Alekseyevich; KOLTUNOVA,

W.P., red.; BCBROVA, Ye.M., tekin. red.

[Business accounting for locomotive repair plants; practices of
the V.I. Lenin Locomotive Repair Flant in Rostov] Eloziaistvennyi
raschet na parovozoremontnom zavode; iz opyta raboty Rostovskogo
parovozoremontnogo zavoda im. V.I. Lenina, 1958. 101 p.

(Rostov-on-Bon—Locomotives—Maintenance and repair)

(Rostov-on-Bon—Locomotives—Maintenance and repair)

GALKIN, M.A.

25(5)

PHASE I BOOK EXPLOITATION SOV/2934

- Burmistrov, Nikolay Semenovich, (Deceased), <u>Mikhail Aleksandrovich</u>
 <u>Galkin</u>, Pavel Fedorovich Matveyev, Grigoriy Akimovich Neshitov,
 and Nikolay Georgiyevich Ozhimkov
- Planirovaniye vspomegatel'nykh tsekhov mashinostroitel'nogo zavoda (Planning the Setup of Auxiliary Shops at a Machine-Building Plant) 2nd ed. Moscow, Mashgiz, 1958. 278 p. 4,000 copies printed.
- Ed.: N.S. Burmistrov, Engineer (Deceased); Reviewers: B.V.
 Voskresenskiy, Economist; P.G. Kalinin, Economist; and A.I.
 Shuster, Economist; Ed. of Publishing House: A.A. Salyanskiy;
 Tech. Ed.: V.D. El'kind; Managing Ed. for Literature on
 the Economics and Organization of Production: T.D. Saksaganskiy.
- PURPOSE: This book is intended for employees at machine-building plants who are engaged in planning.
- COVERAGE: The book deals with problems in planning the setup and operations of various auxiliary shops and services at a

Card 1/7

Planning the Setup (Cont.)

SOV/2934

machine-building plant. The organization of work in such auxiliary units as the machine-repair shop, the tool shop, the industrial power plant, the transportation service, etc. is reviewed, and suggestions are made for improving their labor productivity. Production and maintenance costs of auxiliary shops and units are analyzed, and possibilities of reducing cost investigated. Preparation of estimated expenditures and of monthly financial statements showing results of operations are discussed. The operation of each auxiliary shop or service of the plant is analyzed. Several chapters are written by different authors. No personalities are mentioned. No references are given.

TABLE OF CONTENTS:

Preface

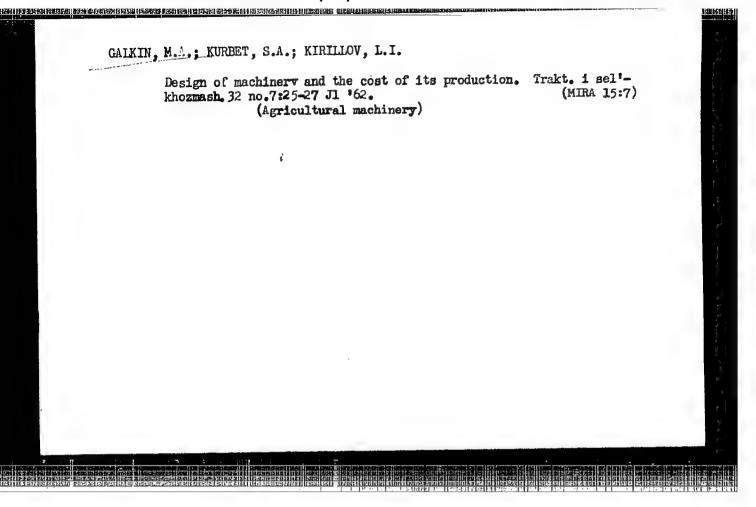
Economic Importance of Auxiliary Shops and Services at a Machine Building Plant (Burmistrov, N.S.) Ch. I.

Significance of auxiliary shops and services in the organization of production

Card 2/7

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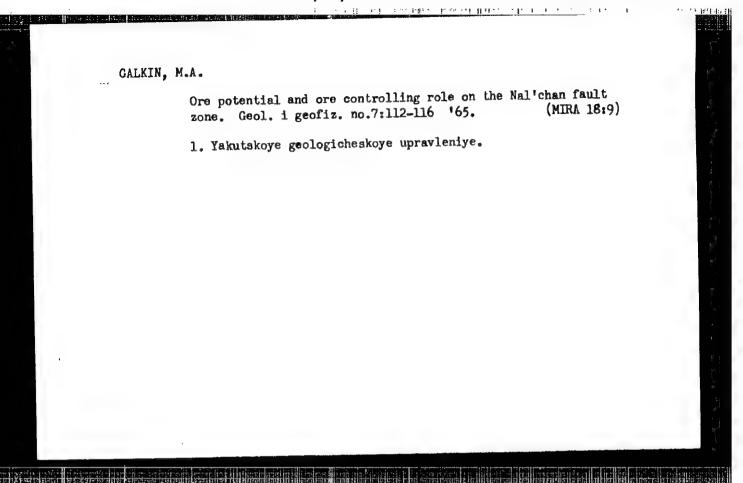
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GALKIN, Mikhail Aleksandrovich; FOPKOV, Ivan Varfolomeyevich;
GURCANOV, B.S., red.; KHODASEVICH, Yu.G., mlad. red.

[Collection of problems for the course "The organization and planning of an industrial enterprise"] Sbornik zadach po kursu "Organizatsiia i planirovanie promyshlennogo predpriiatiia." Moskva, Ekonomika, 1965. 135 p.

(NIRA 18:5)



GALKIN, Mikhail Fedorovich; SOLONIN, Anatoliy Nikolayevich; SANEOMIRSKIY, Mark Moiseyevich; SHAKHOV, Mikhail Alekseyevich; ZHERMUNSKAYA, L.B., inzh., red.; FREGER, D.P., red.izd-va; BELOGUROVA, I.A., tekhn. red.

[Nickel-free 5KhGV steel for forging dies] Beznikelevaia stal'
5KhGV dlie shtampov pri goriachei shtampovke. Leningrad, 1961.
14 p. (Leningradskii Dom nauchno-tekhnicheskoi propagandy. Obmen
peredovym opytom. Seriia: Metallovedenie i termicheskaia obrabotka, no.7)
(MIRA 14:12)
(Steel alloys—Testing) (Dies (Metalworking))

SOLNTSEV, Yuriy Parfir'yevich; GALKIN, Mikhail Fedorovich; LITVAK, Valeriy Abramovich; SLITSKAYA, I.M., inzh., red.; SHILLING, V.A., red. izd-va; BELOGUROVA, I.A., tekhn. red.

[Reducing metal consumption for risers of ingots and castings]
Puti snizheniia raskhoda metalla na pribyl'nuiu chast' slitkov i
otlivok. Leningrad, 1961. 21 p. (Leningradskii Dom nauchnotekhnicheskoi propagandy. Obmen peredovym opytom. Seriia: Liteinoe
proizvodstvo, no.2)

(MIRA 14:7)

GALKIN, Mikhail Fedorovich; SOLMITSEV, Yuriy Porfir'yevich; SERCV,
Gennadiy Vladimirovich; SOKOLOV, A.N., red.; KATSEEL'SON,
N.Ye., red.izd-va; GVIRTS, V.L., tekhn. red.

[Improved procedure for the smelting of lKhl8N9TL steel]
Usovershenstvovanie tekhnologii vyplavki stali lKhl8N9TL
Leningrad, 1962. 20 p. (Leningradskii dom nauchnotekhnicheskoi propagandy. Obmen peredovym opytom. Seriia:
Liteinoe proizvodstvo, no.4) (MIRA 15:10)
(Chromium-nickel steel--Metallurgy)

EWT(d)/EWT(m)/EWP(c)/EWP(v)/T/EWP(t)/EWP(k)/EWP(h)/EWP(1)UR/0137/65/000/012/10046/1046 ACC NRI SOURCE CODE: AR6009951 AUTHORS: Vladimirov, N. F.; Galkin, M. F.; Sointsev, Yu. P. TITLE: Development of programmed electrical operating conditions for the smelt of steel in an arc furnace in connection with automation of the process SOURCE: Ref. zh. Metallurgiya, Abs. 12V347 REF SOURCE: Elektrotermiya. Nauchno-tekhn. sb., vyp. 44, 1965, 64-67 TOPIC TAGS: steel, steel industry, arc furnace, smelting furnace, computer programming, computer/ VU-5086 computer ABSTRACT: A technique is presented for formulating a computer program based on the electrical operating conditions corresponding to the most economical conversion of 1 ton of steel. On the basis of the characteristics of 400 smeltings of various steel types in a furnace of 3-ton nominal capacity, the correlation dependences of the furnace operation characteristics -- the specific smelting period T and the specific electrical power consumption W -- on the mean active power P were found in the form $\tau = A_1 + B_1 P + C_1 P^1$ $W = A_s + B_s P + C_s P^s$ where A1, B1, C1, A2, B2, and C2 are the coefficients of the regression equation. Calculations of the most economical power permit values to be determined for the duration and mean active power for particular stages of the smelting period: 669.187:621.365.2 UDC:

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1) $t_1 = 8 \text{ min}$, $P_1 = 1360 \text{ kw}$; 2) $t_2 = 40 \text{ min}$, $P_2 = 1750 \text{ kw}$; 3) $t_3 = 15 \text{ min}$, $P_3 = 1500 \text{ min}$								
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621.746.757:669.14.018 8

AUTHOR: Gulyayev, B. B. (Doctor of technical sciences); Kursin, A. V. (Engineer);
Galkin, M. F. (Candidate of technical sciences); Chiviksin, Ya. Ye. (Engineer)

TITLS: Defects in high-alloy steel castings and their prevention

SOURCE: Liteynoye proisvodstvo, no. 7, 1965, 1-3

TOPIC TAGS: high alloy steel, casting defect, steel casting, chromium steel

ABSTRACT: The development of chemical machine building has magassitated extensive casting of complex alloy steels containing 18-207, of Cr. About 0, 10% C,
N1, km, T1, and other admixtures, These steels are hard to handle and the intensive interactions of Cr. T1, and Mn with atmospheric oxygen and nitrogen during smelting and casting result in specific casting defects - blisters and subskin porousness. Even minute variations in the content of the basic components (affect the mechanical and other (corrosion resistance) magnatic permeability)
properties, Orhe paper describes a detailed study of the nature of these defects in Kh18N9TL, Kh20N5G12AFL and Kh25N5TMFL steels, and offers detailed recommendations for and description of casting produces which prevent the appearance of Card 1/2

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GALKIN, M.I., inzhener.

Experience from the construction of the Leningrad Reinforced Concrete Products Factory. Biul.stroi.tekh. 13 no.2:7-11 F '56.

(MLRA 9:5)

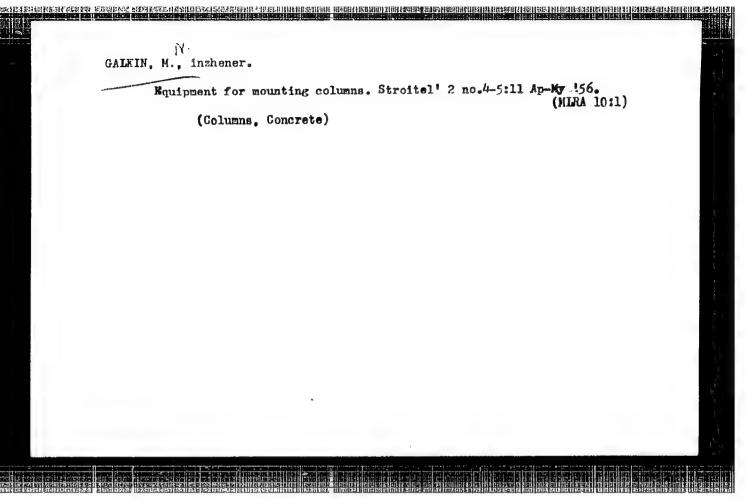
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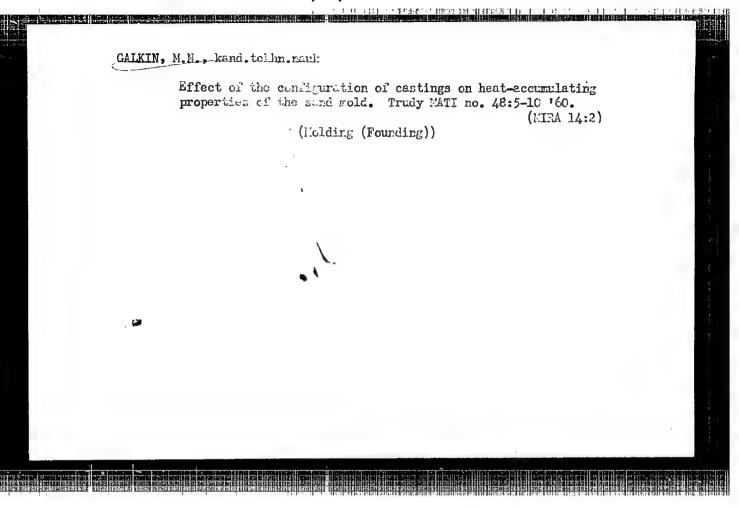
(Reinforced concrete construction)

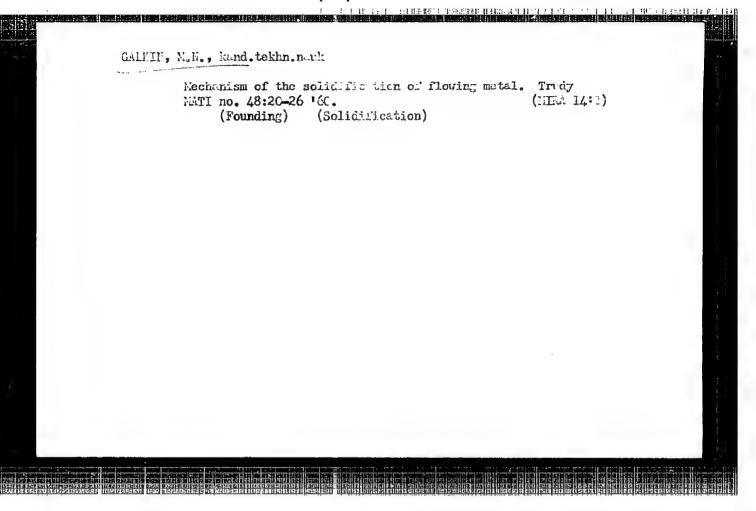
GALKIN, M. N:

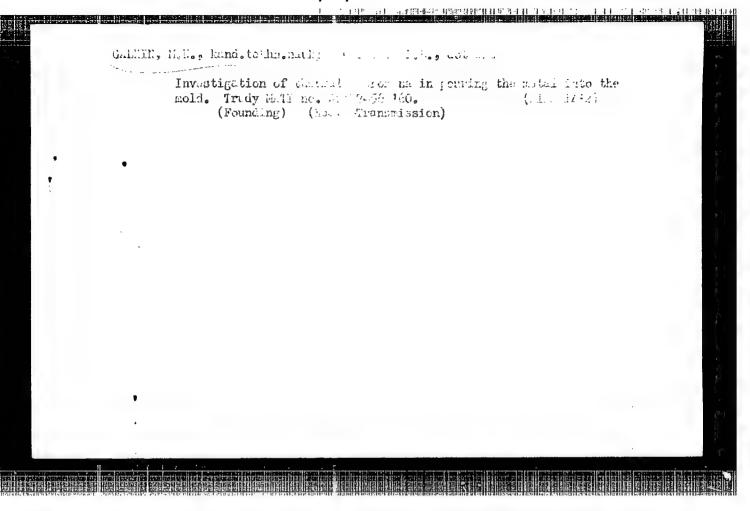
GALKIN, M. N.: "The thermal consitions for the flow of metal in the channels of a casting mold." Min Higher Education USSR. Moscow Aviation Technological Inst. Moscow, 1956.

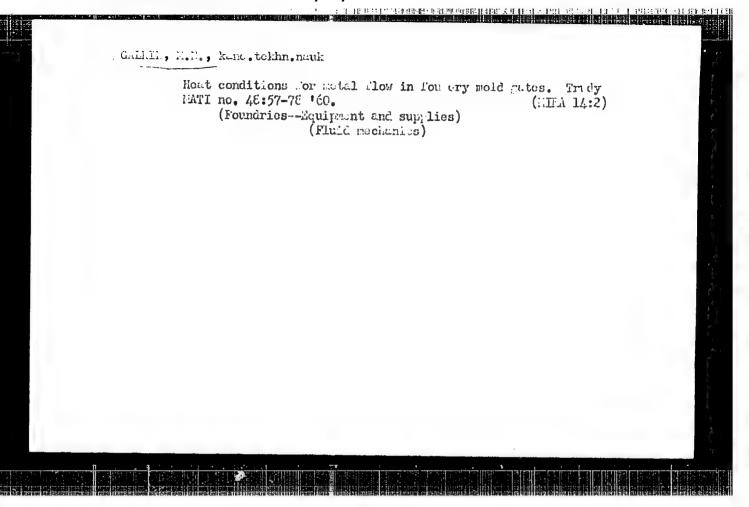
SO: Knizhnaya letopis'
No 21, 1956. Moscow

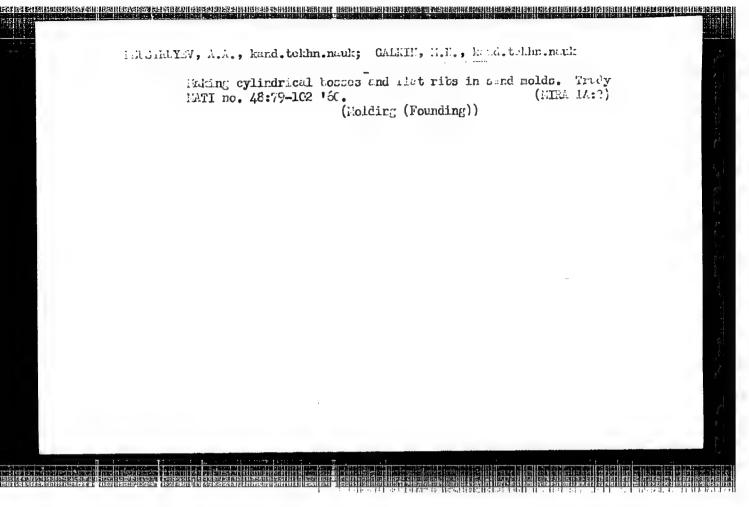












ACCESSION NR: AT4019715 \$/2536/63/000/058/0005/0020 AUTHOR: Galkin, M. N. (Candidate of technical sciences, Docent); Stebalov, Ye. S. (Candidate of technical sciences) TITLE: Squeeze casting of thin-walled panels SOURCE: Moscow. Aviats. tekhn. institut. Trudy*, no. 58, 1963. Teploobmen pri litt. ye vy*zhimaniyem (Heat exchange during squeeze casting), 5-20 TOPIC TAGS: squeeze casting, squeeze casting technique, squeeze casting unit LV-1, thin-wailed panel, squeeze cast panel, aircraft construction ABSTRACT: The author's analyze the process of filling stationary casting molds and discuss squeeze casting mechanisms for an angular or plane-parallel return of the matrix from one or both sides. A description is given of a currently operational unit LV-I (see Figs. I and 2 in the Enclosure). The unit's weight is 9500 kg net; it is 3240 mm long, 2850 mm wide and 1500 mm high. Maximum lid return pressure is 6800 kg. The unit is operated by two men and can produce 4-6 casts per hour (up to 2100 mm long, 1200 mm high and 1 mm or more in thickness). Preparation of the unit and the operating procedure are described. Tolerances are held to $\pm~0.5~\mathrm{mm}$ on panel face across a length of 200 mm, + 0.3 mm for thickness of wall and ribs and + 1 mm for linear dimensions and diameters of lugs. The microvariance of the

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fa. 3%	face surface does not exceed 0.04 mm. Average elongation of the cast panels was 3%, tensile strength 22 kg/mm². Orig. art. has: 15 illustrations and I graph.													
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ACCESSION NR: AT4019716

\$/2536/63/000/058/0021/0046

AUTHOR: Galkin, M. N. (Candidate of technical sciences, Docent); Govseyev, L. L. (Docent)

TITLE: Thermal analysis of a casting crucible

SOURCE: Moscow. Aviats. tekhn. institut. Trudy*, no. 58, 1963. Teploobmen pri lit'ye vy*zhimaniyem (Heat transfer during squeeze casting), 21-46

TOPIC TAGS: squeeze casting, casting, crucible, crucible design, temperature gradient, alloy casting, alloy temperature, casting temperature, alloy hardening

ABSTRACT: The flow, cooling and hardening of alloys can readily be regulated during squeeze casting, but thin-walled castings of high quality can only be obtained with strict regulation of the thermal and hydrodynamic conditions. During casting, the crucible cools rapidly and marked temperature gradients arise in the alloy, which are equilibrated during solidification. For this reason, in the design and construction of crucibles for squeeze casting equipment, special requirements with regard to the temperature field of the alloy should be taken into consideration. The present paper deals with the results of experimental and mathematical studies on the cooling of the alloy during pouring, the temperature field

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in the alloy prior to extrusion from the crucible, and the profile of the solid alloy crust formed on the walls. Two harmonious solutions are derived which permit calculation of the tridimensional temperature field in the alloy during cooling in the ladle and metal conduits, the time of onset of hardening in the crucible, and the amount of crust at any point on the crucible wall. Calculations show that the temperature of the alloy in the crucible drops by 44C during casting, producing a longitudinal temperature gradient of as much as 56C. During the next 50 seconds, the average temperature drops by COC, while the temperature gradient remains unchanged for 33 seconds and then drops to 150 during the last 17 seconds. A hard alloy crust, 2 mm thick, is formed at the ends of the crucible, and the vertical temperature gradient at the center before extrusion can reach 45C. Comparison of the theoretical results with experimental data on the LY-I squeeze casting machine, the crucible of which is shown in the Enclosure, indicates that this approach permits calculation of the optimal temperature field in the alloy prior to extrusion and application of the appropriate corrections in the selection of machine design and thermal parameters. Orig. art. has: 20 figures and 16 formulas.

ASSOCIATION: Aviats. tekhn. institut, Moscow (Institute of Aviation Technology)

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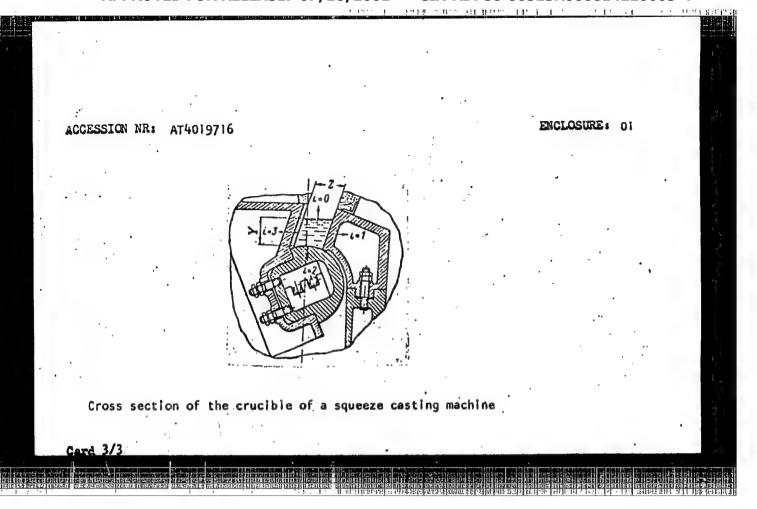
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ACCESSION NR: AT4019717

\$/2536/63/000/058/0047/0080

AUTHOR: Galkin, M. N. (Candidate of technical sciences, Docent)

TITLE: The theory of alloy squeezing and panel formation

SOURCE: Moscow. Aviats. tekhn. institut. Trudy*, no. 58, 1963. Teploobmen pri lit'ye vy*zhimaniyem (Heat transfer during squeeze casting), 47-80

TOPIC TAGS: squeeze casting theory, squeeze casting technology, metal extrusion, panel formation, squeeze casting, alloy squeeze casting

ABSTRACT: The flow, cooling and hardening of alloys can readily be regulated during squeeze casting, but thin-walled castings of high quality can only be obtained with strict regulation of thermal and hydrodynamic conditions. In the present paper, the author describes the mathematical theory of alloy squeeze casting and panel formation in squeeze-casting machines with a plane-parallel and angular (unilateral or bilateral) approach of the dies; panel formation on the LV-1 machine is taken as an example. The theoretical calculations show that it is possible to select the hydrodynamic conditions of alloy extrusion so as to obtain specified relationships between time and the duration, speed and acceleration of alloy motion. Using the formula:

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$$\overline{V} = \frac{\overline{\tau}^{k} - \overline{\tau}^{mk}}{1 - \frac{1}{m}}$$

the operating conditions of the LV-1 machine may be chosen in consideration of the relationship between time and the angle, speed and acceleration of die rotation. Thus, for example, the geometrical relationship between the angle of approach of the die (Φ) and the position of the alloy front (γ) is given by:

$$\overline{\varphi} = 1 - \frac{1}{\overline{Y} \cdot \overline{Y}_{n} - 1} - \frac{\xi}{\overline{Y}_{0}} - \frac{\xi}{R_{0}}$$

$$\frac{a}{\overline{Y}_{0}} + 1 + 1 \cdot \frac{1 - \xi}{R_{0}}$$

The alloy cooling and hardening processes during and after extrusion may be cal-

$$\frac{dt - \frac{0}{c} d\overline{g}}{a_1 b_1 + a_2 b_2} = -\frac{\tau_n}{c \gamma} \frac{d\overline{\tau}}{P},$$

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ACCESSION NR: AT4019717

both for the central part and for ribs, nodes, etc. A diagram of the temperature distribution is shown in the Enclosure. It has been shown that the procedures selected for the LV-I machine are close to optimal. However, some improvement in the quality and stability of the castings may be attained by decelerating alloy extrusion in the deadhead so that 40-60% more of the alloy will harden in the working part of the mold. Orig. art. has: 31 figures and 42 formulas.

ASSOCIATION: Aviats, tekhn, institut, Moscow (Institute of Aviation Technology)

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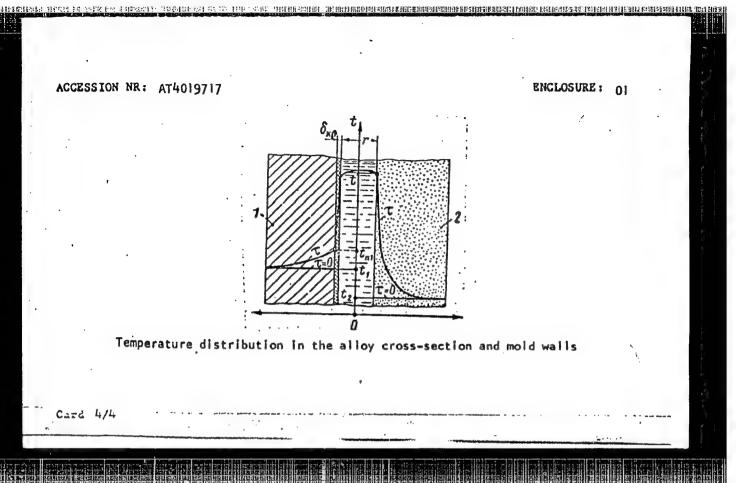
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Card 3/4



BR

ACCESSION NR: AT4019718

\$/2536/63/000/058/0081/0099

AUTHOR: Galkin, M. N. (Candidate of technical sciences, Docent); Tarasutin, T. G. (Engineer); Pushkin, I. L. (Engineer)

TITLE: Thermophysical properties of materials

SOURCE: Moscow. Aviats. tekhn. Institut. Trudy*, no. 58, 1963. Teploobmen pri lit'ye vy*zhimaniyem (Heat transfer during squeeze casting), 81-99

TOPIC TAGS: casting, squeeze casting, steel casting, heat conduction, thermal conductivity, core material, mold wash, core parameter, thermophysical property

ABSTRACT: The flow, cooling and hardening of alloys can readily be regulated during squeeze casting, but thin-walled castings of high quality can only be obtained with strict regulation of thermal and hydrodynamic conditions. In the present paper, assuming that the core material is homogeneous, the authors present a simple experimental technique for determining the principal thermophysical constants of cores and mold washes, as well as the heat capacity and latent heat or solidification of alloys. By the method of pouring metal into molds, the authors derive a relationship between the principal parameters of a and b cores, prepared from wet sand, self-hardening and quick-drying materials, and their density and moisture content or the concentration of binder. The experimental data are shown

ACCESSION NR: AT4019718

in the form of nomograms which permit rapid selection of the appropriate coefficients. In the same way, the authors investigated the thermal conductivity of various washes and the relationship between this value and the number of casting operations. It was found that the thermal conductivity increases with the number of castings, rising particularly sharply after the first one. This increase in thermal conductivity is the result of both an increase in heat conduction and a decrease in thickness. This technique for the experimental determination of the true and average heat capacity and latent heat of solidification simplifies practical tests and increases their accuracy. Orig. art. has: 15 figures, 2

ASSOCIATION: Aviats. tekhn. inst., Moscow (Institute of Aviation Technology)

SUBMITTED: 00

DATE ACQ: 23Mar64

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ACC NRI AT7003186

(A)

SOURCE CODE: UR/2536/66/000/067/0135/0158

AUTHORS: Galkin, M. N. (Doctor of technical sciences, Professor); Kats, E. L. (Engineer)

ORG: none

TITLE: Peculiarities of forming thin-wal

thin-walled hermetic castings

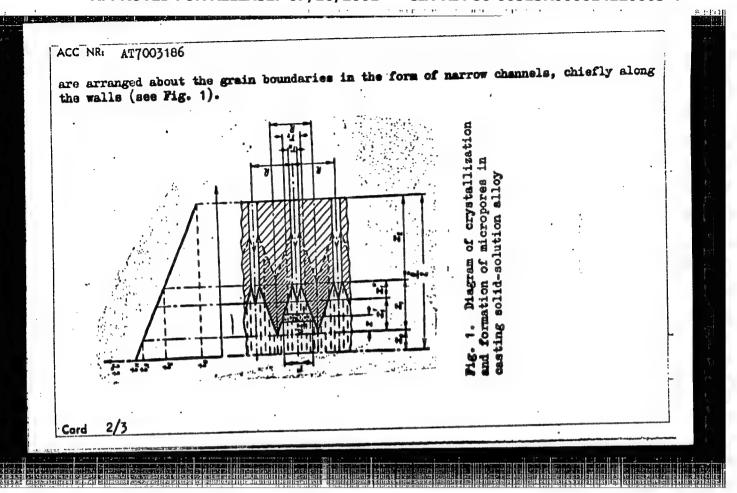
SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 67, 1966.
Voprosy proizvodstva otlivek is alyuminiyevykh splavov (Problems of producing aluminum alloy castings), 135-158

TOPIC TAGS: metal casting, cooling, porosity, temperature distribution, solid solution, aluminum alloy, heat conductivity, grain size/ AL2 aluminum alloy, AL4 aluminum alloy

ABSTRACT: The characteristics of forming thin-walled hermetic castings are examined. The discussion involves the conditions for cooling the melt in the mold, the mechanism of micropore formation and its mathematical description, the feed elements of the casting, and the temperature and solid-phase distributions in the melt when filling the mold. The forming of a two-flange fitting is analyzed. It is found that, from the moment of contact, aluminum alloys are cooled relatively slowly in sand and in painted metal molds. When this low relative rate is changed by one order of magnitude, the microparture of the casting is changed considerably. The micropores

Card 1/3

UDC: 669.714:621.74



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A standard working formula is derived for small values of the relative porosity coefficient as:

 $K = \frac{6 \operatorname{val} M \left(1 + \frac{ca}{0 \cdot 2}\right)}{R_{00} \cdot R^{2} \operatorname{vo}}$

where \triangle t is the hardening time; \mathcal{V} - the kinematic modulus of viscosity; g - the acceleration of gravity; λ - the heat conductivity of the casting material; α_m - the heat conductivity of the mold; \mathcal{V}_m - the temperature drop in the mold; \mathcal{C} - the latent heat of solidification; R - the macrograin size; γ - the density of the melt; and p_0 - pressure. Alloys with a high cutectic content have areas with a flat hardening front. Equations are derived for the temperature field and the solid-phase distribution that make it possible to analyze the forming of a large thin-walled casting from the moment of pouring. Orig. art. has: 18 formulas, 6 photographs, 9 diagrams, and 3 graphs.

SUB CODE: 13/ SUBM DATE: none

Cord 3/3

GARANG NESS	
	1978. Galkin, M. S., On the solution of the Cauchy problem for secretain equation (in Russian), Prikl. Mat. Mold. 20, 2, 271-278,
	MarApt. 1956. Author considers solution of the following initial-value problem.
	$[au'''(x)]^{n'} = bu(x) = 0$, in the interval $(0, 1)$, where a and b are positive function; if x ,
	which together with their first derivatives are continuous in the given interval. This kind of differential equation controls trans- verse vibrations of beams. In general the problem has to be
	solved by some approximate medicd. If the solution happens to be oscillatory then an approximation would very likely choicals large errors. Author makes a rigorous sasiyais of the generation of
	errors in such a case. He shows that when the functions if v and
	exp(x) and exp(-x). He also shiws how to obtain these perion and how to exclude them from the solution. The results are cinicalord in three theorems with long and quite involved proofs.
	T. Lasti, USA
	Here expenses the second control of the seco

Name: GALKIN, M. S.

JPRS/DC-281
CSO DC-1906

Dissertation: Methods for computing proper natural oscillations in a case of close natural frequencies

Degree: Cand Phys-Math Sci

Lefended dissertation: Acad Sci USSR, Mathematical Inst imeni V. A. Steklov

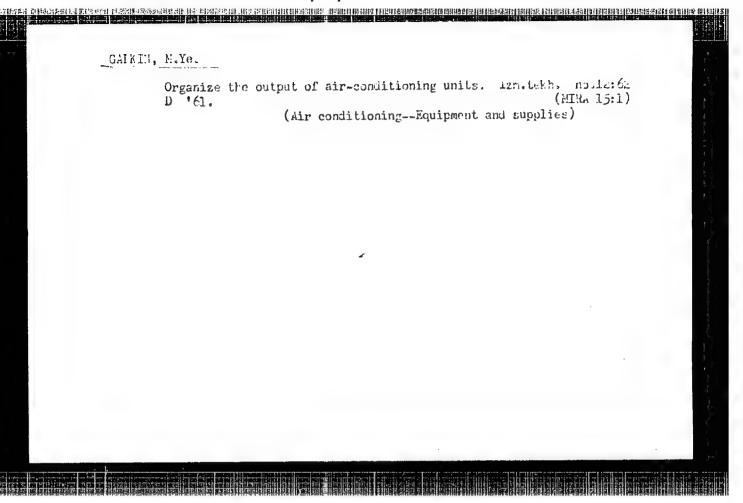
Cublication: 1956, Moscow

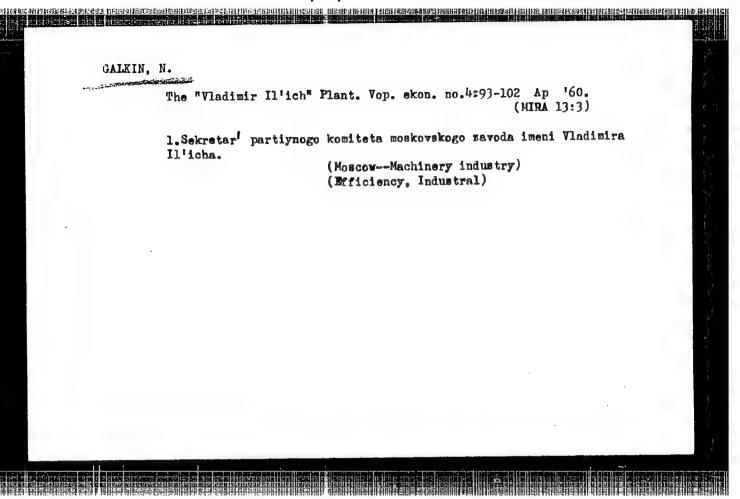
Source: Knizhnaya Letopis', No 2, 1957

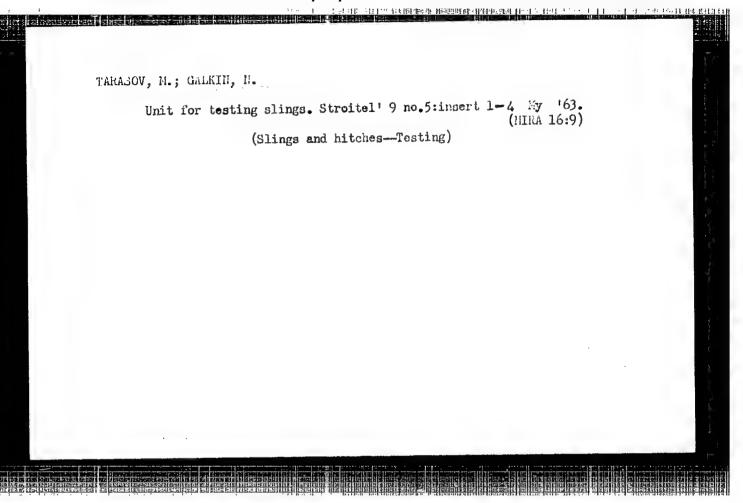
SMELOVSKIY, N.V.; GALKIN, M.Ye.

Experience of the First State Bearing Plant in introducing spring measuring heads. Ism. tekh. no. 1:10-11 Ja '61.

(Measuring instruments)







Girlady, f., polkovník, voyennyy letchik privogo klasovillydie, s., podpolkovník, voyennyy letchik vterego klasovillydie, voyemyy shturman pervogo klasovillydie, voyemyy shturman pervogo klasovillydie, and i kosm. 48 no.12: 53-56 D *65.

EWT(d)/EWT(m)/FA/EWP(h)/EWP(1) Po-4/Po-4/Po-1/Pk-4/P1-4 IJP (c) ACCESSION NR: AP5014816 UR/0209/65/000/006/0059/0061 AUTHOR: Glazkov, I. (Colonel, Military pilot first class); Galling (Lieutenant Colonel, Military pilot second class) TITLE: Automatic longitudinal balancing system for turbourop aircraft SOURCE: Aviatsiya i kosmonavtika, no. 6, 1965, 59-61 TOPIC TAGS: aircraft control equipment, turboprop aircraft ABSTRACT: Among the various stabilizers installed in aircraft, automatic longifudinal balancing systems have lately been used on a large scale. An automatic altitude-control trimmer can be employed to extend the use of the autopilo. improve maneuverability, and increase flight safety while flying with the autopilot turned on. This control system, which automatically provides for the longitudinal balancing of an aircraft guided by autopilot, consists of force sensors, angular velocity transducers, critical trimmer-deflection sensors, trimming and time delay units, relay amplifiers, indicator lamps, and a control button. Card 1/3

L 58474-65-

ACCESSION NR: AP5014816

The sutomatic control system does not exclude the use of a semiautomatic push-button control mechanism. Thus, when the pilot is warned by a signal light that the automatic control system has been cut off by the trimmer, he can, without turning off the autopilot, relieve the forces developing on the altitude control system by pressing the "trimmer" button located on the control stick.

It is very easy to control an aircraft by the use of an autopilot with the automatic altitude-control trimmer turned on. However, during a sudden change in flight direction the automatic trimmer may cause an overload. This is also possible in the case of an autopilot malfunction.

To eliminate these safety hazards, the automatic trimmer-control is provided with a device for turning off the system for the duration of evolutions which give rise to angular velocities with respect to lateral axis in excess of 0, 4°/sec, which corresponds roughly to an overload of 0, 1.

The angular velocity transducer; a sensor element, cuts off the system. It generates and converts an electric signal proportional to angular velocity.

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. 9742-66 EWI (d) E L 9742+66 SOURGE CODE: UR/0209/65/000/011/0056/0060 44,55 AUTHOR: Glazkov, I. (Colonel, Military pilot first class); Galkin, N. (Lie (Lieutenant colonel, Military pilot second class); Krylov, V. (Colonel, Military navigator first class) 44.55 ORG: None TITLE: An automatic control system. SOURCE: Aviatsiya i kosmonavtika, no. 11, 1965, 56-60 TOPIC TAGS: aircraft control. system, automatic control system, airborne computer, navigation computer, aircraft autopilot, automatic navigutor, ranigation equipment ABSTRACT: Aircraft guidance control systems not only carry out the functions of automatic control but also issue instructions to the pilot according to which he may perform flight maneuvers according to a prescribed trajectory. The authors describe a "Privod" piloting-navigation system. In addition to an automatic control systems the Privod is coupled with a computer, the radiotechnical equipment of an SP-502 landing system, an RSBN-2 short-range navigation and landing system, and an automatic pilot. It is intended for affecting the landing approach maneuver, Card 1/2

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flight on a prescr of the system, its authors noted that	glide equipment; and for plottin ribed course, trajectory, and al s function, and landbased aids s t the Privod system presents no	ltitude. The ruch as beac o difficulty fo	various (ons, are (or the cre	compone liscusse w in flig	nts d. The
The experiences	riences of the authors may prov of the authors are to be related				
has: 3 figures.					
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D. 15404-66 EWT(d) IJP(c) BC

ACC NR: AP6000628

SOURCE CODE: UR/0209/65/000/012/0053/0056

AUTHOR: Glazkov, I. (Colonel, military pilot first class); Galkin, N. (Lt. Colonel, military pilot second class); Krylov, V. (Colonel, military navigator first class)

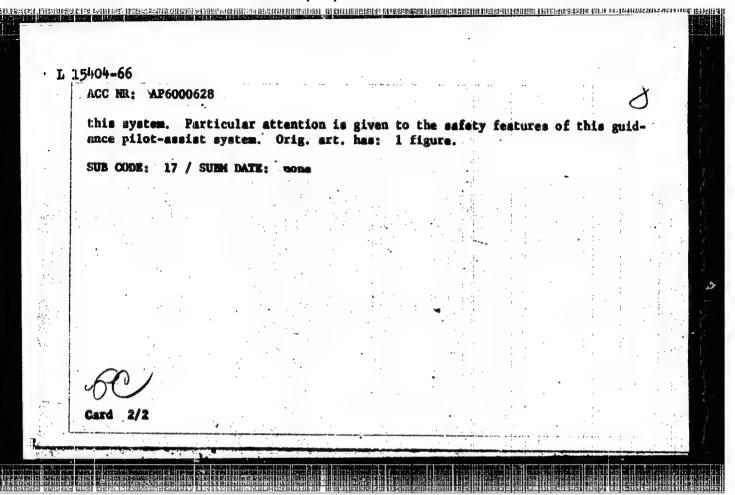
ORG: None

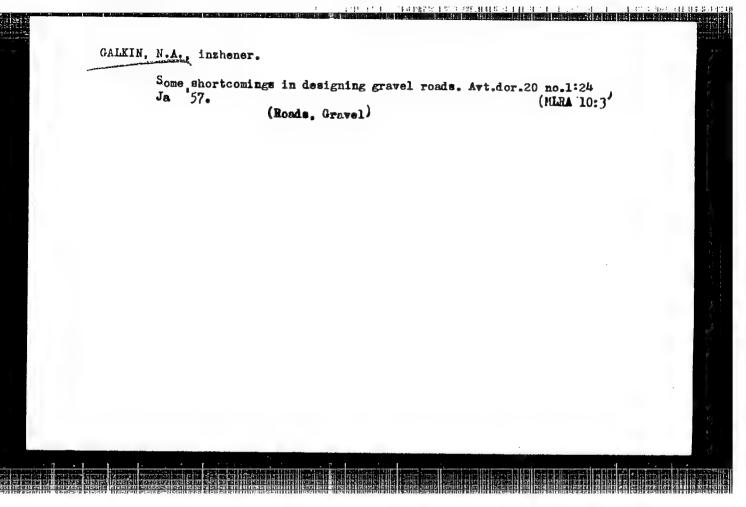
TITLE: The landing approach according to the "Privod" system

SOURCE: Aviatsiya i kosmonavtika, no. 12, 1965, 53-56

TOPIC TAGS: aircraft guidance equipment, command guidance system, aircraft guidance

ABSTRACT: The authors describe the actions of the crew and the piloting technique using the "Privod" command guidance system, both in the landing approach situation and during a flight with a prescribed itinerary (cross-country flight). The authors analyze in some detail landing approach techniques from a square configuration called a "box" as well as the techniques associated with a straight approach run pattern. Wind velocity drift angle corrections and banking angles are analyzed as they pertain to landings based on this system. Three to five training flights are sufficient to enable a good crew to land an aircraft with Card 1/2



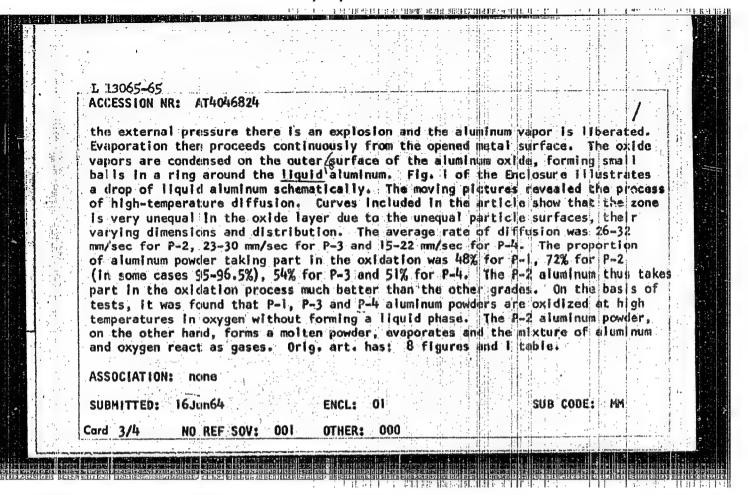


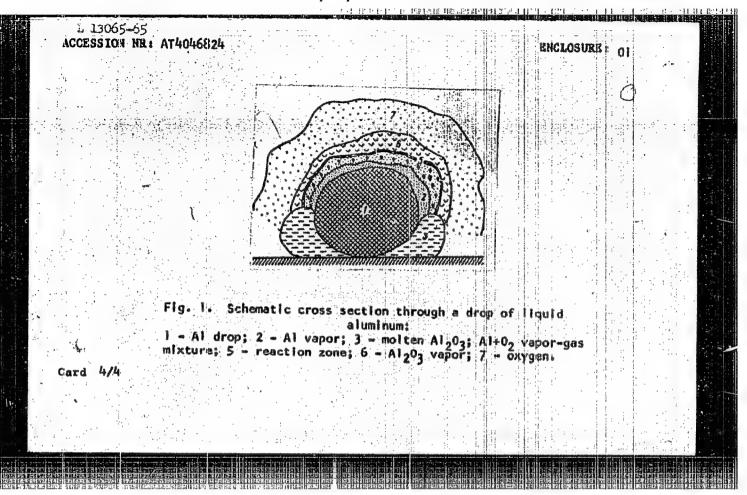
EPA(s)-2/EWT(m)/EPF(n)-2/EWA(d)/EFR/EWP(t)/EMP(b) Ps-4/Pt-10/ Pu-1 IS(mp)-2/ASD(m)-3/BSD/SSD(a) MJW/JD/WW/JG/WB/MLH \$/0000/64/000/000/0096/0103 ACCESSION NR: AT4046824 AUTHOR: Korneyev, V. L.; Vernidub, I. I.; Galkin, N. F.; Dobrokhotov, L. N.; Gostev, Ye. A. TITLE: High temperature oxidation of aluminum powder N SOURCE: AN SSSR. Nauchny'y sovet po probleme zharoprochny*kh splavov. Issledovaniya staley I zplavov (Studies on steels and alloys). Moscow, Izd-vo Nauka, 1964, 95-103 TOPIC TAGS: aluminum powder, aluminum powder oxidation, high temperature oxidation ABSTRACT: Considerable attention is currently being paid to high-temperature metal oxidation. The present article is a continuation of investigations (see V. L. Korneyev and I. I. Vernidub Vy*sokotemperaturnoye okisleniye dispersnogo alyuminiya. Sb. "Issledovaniya po zharoprochnyam splavani, vol. 7. Izdave Ah SSSR, 1961) on the high-temperature exidation of aluminum, including the results of a further study of the process of high-temperature pridation of aluminum powder in oxygen. Standard aluminum powder, grades P-13 P-23 and P-4 with densities of 0.975, 0.825, 1.075 and 0.924, respectively, were used together with bottled oxygen / A special unit designed for the exidation is described in the article.

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ACCESSION NR: AT4046824

The 0.8 g sample was placed in an even layer on a quartz plate. A certain oxygen flow and pressure were then set, the HPO-2 oscillograph was switched on, and the mixture was illuminated intermittently by automatic electric flashes. The completeness of the reaction was found by chemical analysis of the reaction products, and the oxygen consumption was measured on the oscillogram. The entire process was filmed by a special SKS-1 movie camera at 2000-3000 frames per second. It is assumed that a primary oxide film is formed on the surface of P-1, P-2, P-3 and P-4 aluminum powders, insulating the aluminum from direct contact with the oxygen. Therefore, for further exidation, the aluminum and exygen atoms must penetrate through the oxide film. On the basis of tests, it is assumed that the heat from the flame penetrates through the aluminum layer. For highly dispersed aluminum powder, the emitted heat is sufficient for penetration into the aluminum layer. A certain number of aluminum and oxygen atoms penetrate through the oxide film. The reaction causes emission of heat which is used for further heating of the powder, accelerating the reaction, and the process develops at such speed that no liquid phase is formed. The formation of individual spots of motten aluminum is explained by local heat emission sufficient to melt the metal. The tests showed direct formation of a liquid metal phase during high-temperature exidation of P-2 aluminum powder. Further oxidation may cause boiling and evaporation of the 11quid aluminum. The oxide film prevents escape of aluminim vapor into this atmosphere, Most of the vapor therefore remains and when the Internal pressure exceeds





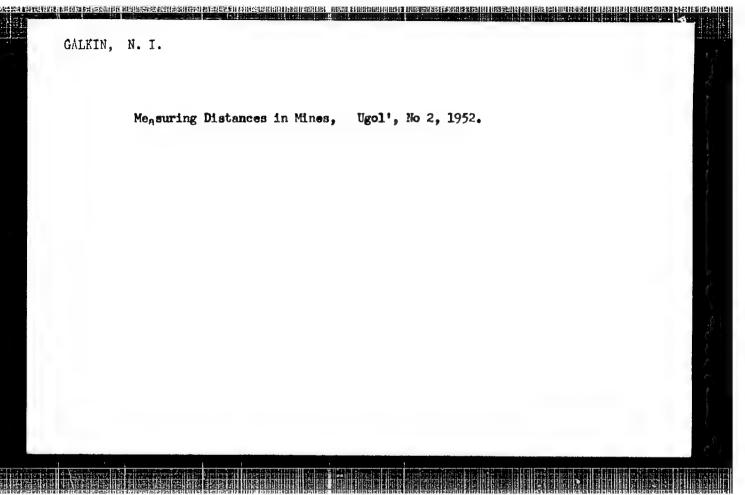
GALKIN, N. I.

TARASOV, P. V. - Inzh. i. KORCHAGIN. A. A. - Inzh., SAKHAROV, I. G. - Avkh., GALKIN. N. I. St. Nauchn., FILLIPOV, A. V. - Chl.-Korr. Akademii Arkhitektury SSSR Prof.

Nauchno-issledovateliskiy institut stroi-telinoy tekhniki Akademii arkhitektury SSSR

Tipy keramicheskikh izdeliy, tekhnologiya ikh izgotovleniya i metody krepleniya Page 100

SO: Collection of Annotations of Scientific Research Work on Construction, completed in 1950, Moscow, 1951



的概则的企业时间不知的证明。这是不是一个人,这是一个人,我们就是一个人,我们是一个人,我们是一个人,我们是一个人,这一个人,这一个人,这一个人,我们是一个人,我们

S/182/61/000/011/003/005 D038/D113

AUTHOR:

Galkin, N. I.

TITLE:

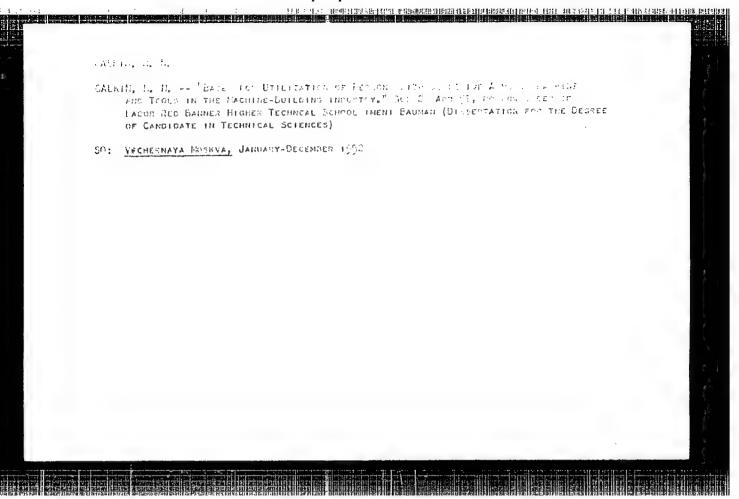
On the precision of forged parts

PERIODICAL: Kuznechno-shtampovochnoye proizvodstvo, no. 11, 1961, 23-25

TEXT: The article deals with erroneous data given in technical literature on the dimensions of formulas for working parts of dies and punches. As an example, the author discusses a book by M. Ye. Zubtsov entitled "Listovaya shtampovka" (Sheet Forging) and published by Mashgiz 1958, and states that dies for extruding, blanking and forging calculated by Zubtsov had to be redesigned, as the diameters in one case were 0.050 mm smaller and 0.400 mm smaller in another. In other cases formulas for punching and extruding holdow workpieces proved to be inaccurate. The author concludes that formulas for calculating the dimensions of the working parts of dies, in the new edition of this otherwise fairly good book, ought to be corrected in accordance with the author's suggestions, and the specifications brought in line with FOCT 7713-55 (GOST 7713-55). There are 2 figures and 2 Soviet-bloc references.

Card 1/1

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	Pub. 61 - 18/23			
Authors :	Galkin, N. M., and Ivanov, Z. F.			
Title :	Machine for cutting skeleton wire			
Periodical :	Lit. proizv. 3, page 29, May-June 1954			
Abstract :	A simple machine for rapid cutting of rods and bars), is described. Drawing	skeleton wire (pro	duction of	
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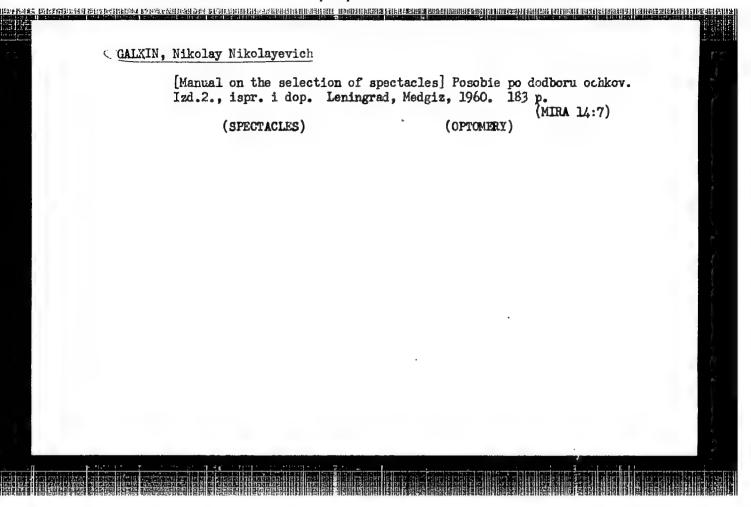
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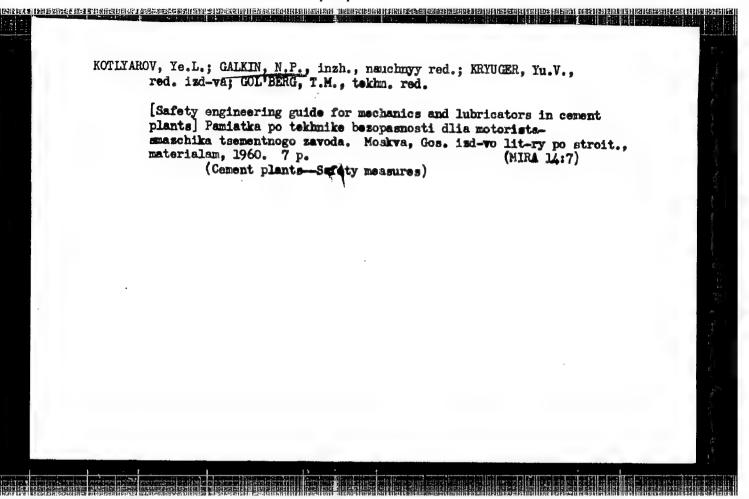
GALKIN, N. N.

"Guide on the Choice of Spectacles" Leningrad, 1955, by the Leningrad Section of the State Publishing House for Medical Literature.

This book contains practical advice for occulists and opticians on the methods of choosing the appropriate spectacles, with basic theoretical principles.

SO: D526604





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等 电分形 人名伊尔 (李明智歌門海海原開院等部門伊里斯海河 4) 体上上层层 。一个月(日):《李氏·司诗宗后 日本巴拉

Galkin, Nikolay Petrovich, and Vladislav Borisovich Tikhomirov

Osnovnyye protsessy i apparaty tekhnologii urana (Principal Processes and Equipment in Uranium Production) Moscow, Gosatomizdat, 1961. 218 p. 5000 copies printed.

Ed. (Title page): B. S. Kolychev, Candidate of Technical Sciences; Ed.: Z. D. Andreyenko; Tech. Ed.: S. M. Popva.

PURPOSE: This book is intended for technical personnel of plants, scientific research institutes, and design bureaus of the uranium industry, and may also be used as a textbook at chemical engineering and mining schools of higher technical education.

COVERAGE: Principal processes and equipment used for recovering uranium from ores are discussed. Concise information on the theory of uranium production processes (grinding, classification, dehydration, leaching, ion exchange, hydrometallurgical extraction, crystallization, drying, and stirring) is presented, and modern

Card I/7

rincipal Processes and Equipment (Cont.)	sov/5823				
methods of calculating these processes are equipment of uranium industry plants is deformance characteristics are given. The at Kolychev for his assistance. References, apany each chapter.	soribed and its pre- uthors thank B. S.				
ABLE OF CONTENTS:					
Foreword 3					
Introduction	5				
PART I. MECHANICAL P	ROCESSES				
Ch. I. Grinding 1. Classification of grinding processes 2. Crushing 3. Fine grinding 4. Crushing and grinding equipment	7 7 8 10 11				
Card 2/7					

MARTIN, F.S.; MAYLS, Dzh.L.[Miles, G.L.]; ZARUBIN, A.I.[translator]; KO-LYCHEV, B.S. [translator]; SAGALOVICH, I.D. [translator] GALKIN, N.P., prof. Wektor tekhn.nauk, red.; KAMAYEVA, O.M., red.izd-va; ATTOPOVICH, M.K., tekhn.red.

[Chemical processing of nuclear fuels] Khimicheskaia pererabotka iadernogo topliva. Moskva, Gos. nauchno-tekhn. izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1961. 264 p. Translated from the English. (MIRA 14:8)

1. Head of Chemistry Section, Australian Atomic Energy Commission (for Mayls).

(Nuclear fuels)

21(1) 5(2)
AUTHORS:

Galkin, N. P., Tikhomirov, V. B., Goryaynov, N. Ye., Fedorov,

V. D.

TITLE: The Mechanism by Which a Liquid Is Dispersed in a Plate Extractor and Ways of Improving the Dispersion (Mekhanizm dispergirovaniya zhidkostey v tarel'chatom ekstraktore i sposob yego intensifikatsii)

PERIODICAL: Atomnaya energiya, 1959, Vol 7, Nr 2, pp 159 - 160 (USSR)

ABSTRACT: The difference between the normal and the better modified version of the extractor consists in the fact that in the modified extractor an air inlet pipe is installed beneath the inlet for the light phase. This opening of the pipe is in the center of the column and is directed upwards. There are no overflow pipes in the extractor. The whole stream has to pass thru the openings in the plate. A stable operation of the column is ensured when the airflow moves at 0.03 m/s over the whole cross section of the column. When the airconsumption increases, bubbles form between the liquid drops and these bubbles reduce the contact surface. The new column with the air agitation system

Card 1/2 incorporated, was tested with the following systems: water -

- TO ROLL 1: 1-12 - 2-2-2-2-2-1-2-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13 | 1-13

The Mechanism by Which a Liquid Is Dispersed in a Plate SOV/89-7-2-9/24 Extractor and Ways of Improving the Dispersion

nitric acid - uranyl nitrate - tributyl phosphate in petroleum. The separation properties are approximately three times higher than those of a normal column. The total liquid load can be $v \not \! 30 \text{ m}^3/\text{m}^2$ in case of an optimum air agitation. The dependency of the extraction capacity upon the intensity of the air agitation was determined by experiment. The result is shown in a diagram. The extraction loss caused by the air stream is negligibly small. There are 2 figures.

SUBMITTED: March 31, 1959

Card 2/2

PHASE I BGOK EXPLOITATION

sov/3830

Galkin, N.P., A.A. Mayorov, and U.D. Veryatin

Tekhnologiya pererabotki kontsentratov urama (Technology of Processing Uranium Concentrates) Moscow, Atomizdat, 1960. 162 p. Errata slip inserted. 4,000 copies printed.

Ed.: T.P. Kalyuzhnaya; Tech. Ed.: Ye. I. Mazel,

PURPOSE: This book is intended for chemical engineers and technicians in uranium production.

COVERAGE: The book presents the theory and description of processes in the treatment of uranium concentrates to obtain pure salts and uranium metal. The authors liscuss the applications of uranium, the properties of uranium and its ions in solution, methods for the production and refining of uranium concentrates, methods for the preparation of uranium tetrafluoride, the preparation of uranium metal, and measures for ensuring the safety of personnel in uranium manufacturing. The author also cites earlier books on uranium by Dah. Kats and Ye. Rabinovich, S.Ye. Bresler, O.A. Songina, and I.P. Kislyakov. There are

Card 1/6

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	l references: 91 Soviet, 101 English, 36 French, 27 Swedish	German, 4 Italian, and
TABLE	OF CONTENTS:	
Prefac	- :e	3
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Ch. II	II. Brief Survey of the Properties of Uranium and i Solution	ts Ions in
1.	Position of Uranium in D.T. Mendeleyev's periodic	
	Electron configuration; atomic and ionic radii of	uranium 14
3.	Isotopic structure	14
	Atomic weight of natural uranium	14
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78330 **so**v/89-8-3-15/32

AUTHORS:

Galkin, N. P., Stepanov, M. A.

TITLE:

Solubility of Uranium Hydroxide (IV) in Caustic Soda.

Letter to the Editor

PERIODICAL:

Atomnaya energiya, 1960, Vol 8, Nr 3, pp 258-261 (USSR)

ABSTRACT:

Little is known about the precipitation of uranium (IV) hydroxide in a strongly alkaline medium. Only recently, Gayer and Leider (see ref) showed that the hydroxide of uranium (IV) is amphoteric. The equilibrium constant of

the reaction:

$$U(OH_{4}) + OH^{-} \iff H_{3}UO_{4}^{-} + H_{2}O$$
 (1)

is 1.7·10⁻⁴. Since the solubility of the hydroxide was studied only up to a 0.6 N concentration of the alkali, the authors decided to check the applicability of the above relation for more concentrated alkaline solutions. Hydroxide of uranium (IV) was precipitated

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Solubility of Uranium Hydroxide (IV) in Caustic Soda. Letter to the Editor

78330 SOV/89-8-3-15/32

from the hydrochloric acid solution by means of a water sclution of caustic soda. The hydrochloric acid solution of uranium (IV) was prepared following prescriptions found in literature. The hydroxide of uranium (IV) was precipitated by adding 30 ml of a 0.34 N solution of caustic soda to 2 ml of the uranium chloride solution; the tightly closed test tube was kept for 6 hr in an air thermostat at 20° C, with continuous stirring of the contents. Decanting the precipitate three times in a pure argon atmosphere with water, the authors achieved considerable purity. A qualitative reaction on chlorine ion using silver nitrate gave a negative result. authors note that the statement found in Gmelins (Handbuch der Anorganischen Chemie, Auflage 8, Hr. 55-Uran und Isotope, Berlin, 1936, S. 100), that potassium and sodium cannot be washed away from uranium (IV) hydroxide, seems to be wrong. Spectral analysis showed the absence of sodium (below 0.01%) when the precipitation was achieved using the caustic soda solution.

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Solubility of Uranium Hydroxide (IV) in Caustic Soda. Letter to the Editor

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This solution was prepared using chemically clean substances. Next, the authors added to the precipitate fixed quantities of alkaline and distilled water. Probes were then mixed in thermostats at 25 + 1°C during from the day of the da then determined. Results are on Fig. A. The authors state that conclusions of Gayer and Leider are valid only up to a 0.5 N concentration. Above this concentration the linear relationship is destroyed, and Eq. (1) is not valid. The decrease in uranium concentration may be explained by salting out by means of sodium ions, if one assumes that a new compound NaH3UO4 1s formed in the pricipitate. Analyzing the solid phase, the authors came to the conclusion that the proposed compound can be stable only in strongly alkaline media, while in the presence of water an hydrolysis starts which can be described by the equation:

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 $^{\text{NaH}_3^{\text{UO}_4} + \text{H}_2^{\text{O}}} \longrightarrow ^{\text{U(OH)}_4} + ^{\text{NaOH}}$

(2)

Solubility of Uranium Hydroxide (IV) in Caustic Soda. Letter to the Editor

78330 sov/89-8-3-15/32

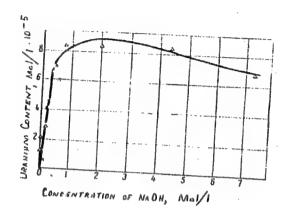


Fig. A. Concentration of uranium (IV) versus alkalinity of medium. (Δ) present data; (x) data by Gayer and

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Solubility of Uranium Hydroxide (IV) in Caustic Soda. Letter to the Editor

78330 SOV/89-8-3-15/32

There are 1 figure; 2 tables and 9 references, 4 Soviet, 2 French, 1 German, 1 Canadian, 1 U.S. The Canadian and U.S. references are: K. Gayer, H. Leider, Canad. J. Chem. 35, Nr 1, 5 (1957); J. Katz, E. Rabinowitz, Chemistry of Uranium, M., Izd-vo inostr. lit., 1954.

SUBMITTED:

November 27, 1959

Card 5/5

84896

S/089/60/008/006/022/023/XX B006/B063

21, 3200 Authors:

Galkin, N. P., Sudarikov, B. N., Zaytsev, V. A.

TITLE:

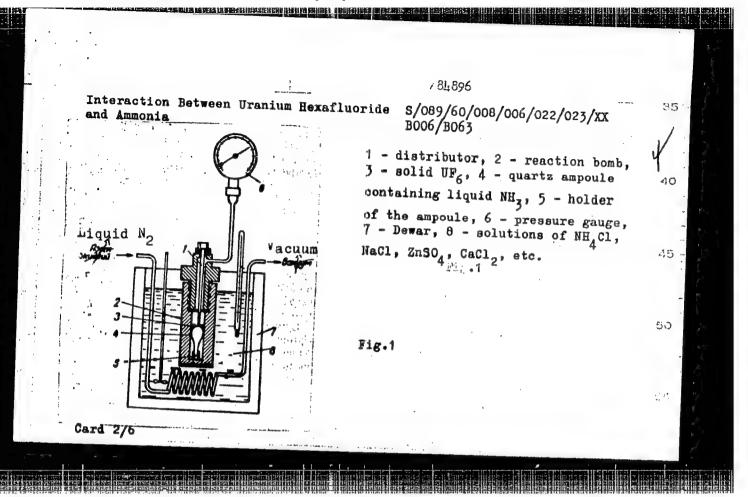
Interaction Between Uranium Hexafluoride and Ammonia

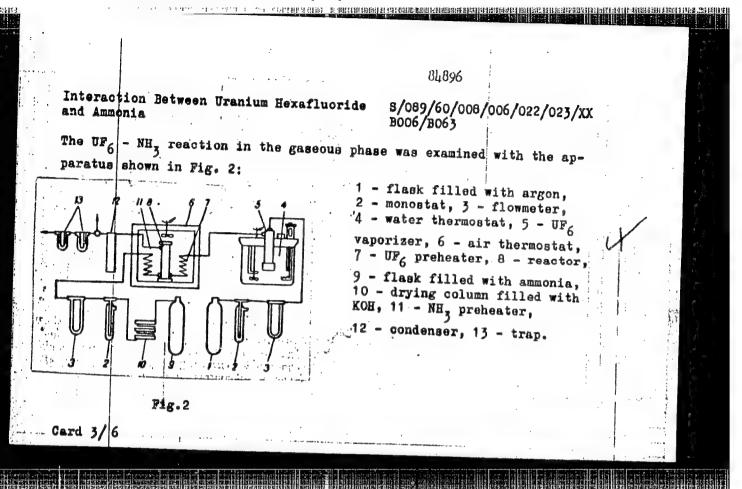
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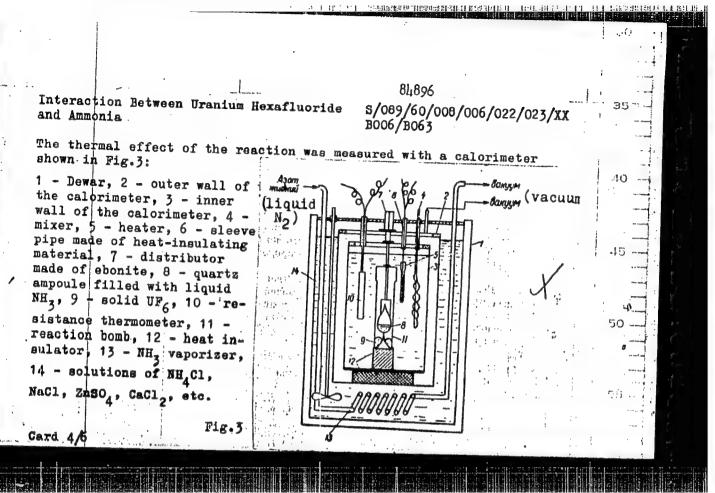
Atomnaya energiya, 1960, Vol. 8, No. 6, pp. 530 - 534

TEXT: The authors studied the interaction between uranium hexafluoride and ammonia in the temperature range from -50 to +200°C for the purpose of determining the reaction equations at different temperatures and measuring the rates and thermal effects of the reactions. The reaction of uranium hexafluoride with liquid and gaseous ammonia was examined with an apparatus schematically shown in Fig.1:

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Table 1	-20 -15 -10 -5	59,5 64,4 71,4 73,8	60,2 58,7 59,8 60,0	32,5 33,9 33,1 32,8	6,9 7,2 7,1 6,8			
	0 +15 +25	77,5 77,6 77,7	62,0 62,2 63,2	30,2 31,0 29,7	6.7	·	X	:
Card 5/6	+100 +150 +200	98,7 99,1 99,5	61,6 62,0 62,2	29,1 29,3 29,0	9,2 8,9 8,7		X	

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Interaction Between Uranium Hexafluoride and Ammonia

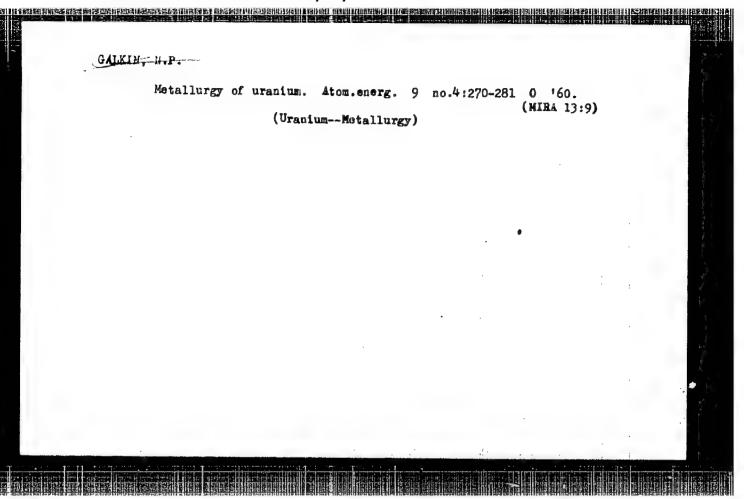
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The entire reaction within the range -50 - -30° C can thus be described by equation $6UF_6$ + $(8+6n)NH_3$ \longrightarrow UF_5nNH_3 + $6NH_4F$ + N_2 , where n = 0.73. The following equations hold in the ranges $0 - +25^{\circ}$ C and $100 - 200^{\circ}$ C, respectively: $4UF_6+8NH_3$ \longrightarrow $2UF_5+2NH_4UF_5+4NH_4F$ + N_2 and $3UF_6+8NH_3$ \longrightarrow $3NH_4UF_5+3NH_4F+N_2$. The calculated values are all compared with the experimental ones. The thermal effect observed between -50 and -30°C varies from 50.8 to 83.6 kcal/mole (cf. Table 2); at -40°C, it coincides with the value calculated from the reaction equation. Within the range -20 to +20°C, the reaction rate was measured as a time function (Fig.4). The functions (-20°, 0°, +20°C) are hyperbolic. There are 4 figures, 5 tables, and 9 references: 3 Soviet, 1 US, 2 German, and

SUBMITTED:

July 15, 1959

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AUTHORS: Step

Stepanov, M. A., Galkin, N. P.

TITLE

The Solubility Product of the Hydroxide of Tetravalent

Uranium A

PERIODICAL: Atomnaya energiya, 1960, Vol. 9, No. 4, pp. 282 - 285

TEXT: The present work gives a calculation of the solubility product of uranium (IV) hydroxide. An exact knowledge of the solubility product is necessary for a rational processing of uranium. In the introduction, the authors discuss results of some related papers (Refs. 1-8). Then, they discuss the determination of experimental data necessary for the calculation. The starting material was a solution in hydrochloric acid of uranium (IV) which was kept in a retort in a pure atmosphere of argon. Even after 15 days no oxidation of the uranium was observed. The concentration was measured titrimetrically with potassium bichromate. It was 0.590 M in relation to uranium and 1.02 M in relation to HCl. Solutions of ammonium hydroxide, sodium hydroxide, and potassium hydroxide (0.464, 1.992, 2.184 N, respectively) were used as precipitants. The pH

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The Solubility Product of the Hydroxide of Tetravalent Uranium

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determination was made with a glass electrode three and eight minutes after the addition of the precipitant. In order to keep the free alkali content in the solution low, the experiment was interrupted when pH attained a value of 7 - 9. The dependence of pH on the amount of the added precipitant is graphically shown (Fig.). It was found that the formation of uranium hydroxide begins in a solution with 5.75·10⁻³g.ion/1 U(IV) and 3.28·10⁻² Cl at pH = 2.38. All pH-measurements were made with a tube potentiometer of the type III-5 (LP-5) which could accurately determine the activity of hydrogen ions up to +0.05 unit. The calculation of the active on the basis of the results of acidimetric titration, led to the value of (1.10 ± 0.72).10⁻⁵² for the solubility product of the uranium (IV) hydroxide. There are 1 figure and 24 references: 20 Soviet, 3 US, and

SUBMITTED: March 18, 1960

Card 2/2

GALKIN, N.P.; PONOMAREV, L.A.; SHISHKOV, Yu.D.; PODOSHVINA, V.A., red.; VLASOVA, N.A., tekhn. red.

> [Plutonium hexafluoride, its preparation and properties] Geksaftorid plutoniia, ego poluchenie i svoistva. Moskva, Gos.izd-vo lit-ry v oblasti atomnoi nauki i tekhniki, 1961. 34 p. (MIRA 15:2)

(Flutonium fluoride)

PHASE I BOOK EXPLOITATION

SOV/5820

N. S. Nikolayev, Yu. D. Shishkov, A. B. Krutikov

Khimiya i tekhnologiya ftoristykh soyedineniy urana (Chemistry and Technology of Uranium Fluoride Compounds) Moscow, Gosatomizdat, 1961. 347 p. Errata slip inserted. 4500 copies printed.

Ed. (Title page): N. P. Galkin, Doctor of Technical Sciences, Professor; Ed.: N. A. Korobtsova; Tech. Ed.: S. M. Popova.

FURPOSE: This book is intended for chemical and nuclear engineers and teachers and students of schools of higher education.

COTERAGE: The monograph reviews Soviet and non-Soviet literature published up to June 1960 on the physicochemical properties of uranium fluorides and methods of producing them from salts, oxides, and metallic uranium. Methods of processing uranium chemical concentrates to the tetra- and hexatluorides, which are initial products in the production of nuclear fuel,

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APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R000614120003-4"

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Chemistry and Technology of Uranium (Cont.) SOV/5820 are of primary interest. Fluoride methods are preferred to hydrometallargical methods because radioactive waste solutions in the former are either reduced to a minimum or eliminated. No personalities are menbioned. References accompany individual chapters. TABLE OF CONTENTS: Foreword 3 Introduction 5 Ch. I. Physicochemical Properties of Uranium Fluoride Compounds 11 Ch. II. Production of Uranium Tetrafluoride From Aqueous Solutions 53 Ch. III. Dry Methods of Producing Uranium Tetrafluoride 78 Ch. IV. Production of Uranium Hexafluoride 136 Card 2/3

GALKIN, N.P.: KAYOROV, A.A.: SEUPIN, V.A.: FOLUELTOVA, G.B.: KRYLOV, A.S.

Composition of precipitates forming in the reaction of ammonia with aqueous solutions of uranyl sulfate or nitrate. Zhur.neorg.khim.
6 no.10:2319-2324 0 '61. (MiRa 14:9)

(Uranyl sulfate) (Uranyl nitrate) (Ammonia)